

## 2025 Spring Semester

### *Syllabus for General Chemistry I-Chemical Principles*

1. **Course:** **General Chemistry I (CH101)** [lecture: Experiment: Credit = 3:0:3]

#### 2. Lecture Timetable

Time (Mondays and Wednesdays)	Class	Professor	Lecture Room(E11)
14:30-16:00	D	Insung S. Choi	202

#### 3. Summary of Lecture

This General Chemistry I lecture offers an overview of fundamental chemistry topics, including a classical approach to chemical bonding, an introduction to quantum mechanics, and a quantum mechanical perspective on atomic and molecular structures. Additional topics are covered to enhance chemical knowledge and deepen understanding of core chemical principles.

#### 4. Material for Teaching:

- o Principles of Modern Chemistry, 8th ed, Oxtoby/Gillis/Campion (Brooks/Cole)
- o Lecture materials will be provided through the KLMS website of each class (<https://klms.kaist.ac.kr/>).

#### 5. General Guidelines

All basic lecture notes can be downloaded at the General Chemistry Website:

<http://www.gencheminkaist.pe.kr> or a link be found at <http://chem.kaist.ac.kr>.

- 1) Practice Sessions led by TAs are scheduled from 8:00 to 8:50 pm on Mondays. These sessions are optional, providing an opportunity for students who seek additional discussion and problem-solving to participate.
- 2) The grading system will be determined based on the total scores achieved by students. The distribution of A grades (including A+, A<sub>o</sub>, and A-) will be less than 50% of the total class. A C+ grade will correspond to a total score of approximately 50 points. Students taking the course with a P/NR grading option must achieve a score higher than 50 points (equivalent to a C+).

## Grading Criteria and Points Distribution (Total 100 points)

I. Mid-term Exam: 34 points    II. Final Exam: 34 points

(Out of 8-10 questions: 1-2 will come from assignments, 1-2 from previous exams)

III. Homework: 16 points

- Chapter summary: 8 points (1 point for each chapter, maximum of 3 pages, only **handwritten assignments** will be accepted, each submission is awarded either 0.5 or 1 point, depending on the content)
- Chapter problem: 8 points (1 point for each chapter, each submission is awarded either 0.5 or 1 point, based on the answers provided)

IV. Attendance & Attitude: 16 points

- Maximum of 16 points (1 point for each attendance of lectures and practice sessions)
- This course does not penalize absences, so there is no recognized attendance.

V. Plagiarism of Homework:

- First instance: Warning with a deduction of 10 points, second instance: F grade

### **6. Waiver Examination**

The waiver examination on General Chemistry I will be held at the beginning of the semester, but only for those, who did not take any previous General Chemistry I classes.

## 7. Lecture Schedule

Week (Mondays, Wednesdays)	Chapters	Topics	Due date for Homework (Chap. Summary & problem)	Practice session (Mon, 20:00 ~ 20:50)	Notes
1	3	Atomic Shells and Classical Models of Chemical Bonding	-	-	Substitute Holiday (03/03)
2	3 and 4	Atomic Shells and Classical Models of Chemical Bonding / Introduction to Quantum Mechanics	-		
3	4	Introduction to Quantum Mechanics	Chap 3		
4	5	Quantum Mechanics and Atomic Structure	Chap 4	O	
5	5	Quantum Mechanics and Atomic Structure	-		
6	6	Quantum Mechanics and Molecular Structure	Chap 5	O	
7	6	Quantum Mechanics and Molecular Structure	Chap 6	O: 4/7(Mon) & 4/11(Fri)	
8	Mid-Term Exam				
9	12	Thermodynamic Processes and Thermochemistry	-	-	
10	12 and 13	Thermodynamic Processes and Thermochemistry / Spontaneous Processes and Thermodynamic Equilibrium	-		
11	13	Spontaneous Processes and Thermodynamic Equilibrium	Chap 12		Holiday (05/05)
12	17	Electrochemistry	Chap 13	O	
13	17	Electrochemistry	-	O	
14	18	Chemical Kinetics	Chap 17	O	
15	18	Chemical Kinetics	Chap 18	O	
16	Final Exam				

## 8. Chapter Problems: To be announced